

List of Common Polyatomic Ions

Positive Ions

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>	<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
NH_4^{+1}	ammonium	+1	Hg_2^{+2}	dimercury (I)	+2

Negative Ions

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>	<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
$\text{C}_2\text{H}_3\text{O}_2^{-1}$	acetate	-1	CO_3^{-2}	carbonate	-2
$\text{C}_7\text{H}_5\text{O}_2^{-1}$	benzoate	-1	CrO_4^{-2}	chromate	-2
HCO_3^{-1}	bicarbonate	-1	$\text{Cr}_2\text{O}_7^{-2}$	dichromate	-2
BrO_3^{-1}	bromate	-1	HPO_4^{-2}	hydrogen phosphate	-2
BrO_2^{-1}	bromite	-1	MoO_4^{-2}	molybdate	-2
BrO^{-1}	hypobromite	-1	$\text{C}_2\text{O}_4^{-2}$	oxalate	-2
ClO_4^{-1}	perchlorate	-1	$\text{C}_8\text{H}_4\text{O}_4^{-2}$	phthalate	-2
ClO_3^{-1}	chlorate	-1	O_2^{-2}	peroxide	-2
ClO_2^{-1}	chlorite	-1	SiO_3^{-2}	silicate	-2
ClO^{-1}	hypochlorite	-1	SO_4^{-2}	sulfate	-2
CN^{-1}	cyanide	-1	SO_3^{-2}	sulfite	-2
OCN^{-1}	cyanate	-1	$\text{S}_2\text{O}_3^{-2}$	thiosulfate	-2
SCN^{-1}	thiocyanate	-1	$\text{C}_4\text{H}_4\text{O}_6^{-2}$	tartrate	-2
CHO_2^{-1}	formate	-1	WO_4^{-2}	tungstate	-2
$\text{C}_2\text{H}_4\text{O}_3^{-1}$	glycolate	-1	AsO_4^{-3}	arsenate	-3
OH^{-1}	hydroxide	-1	AsO_3^{-3}	arsenite	-3
IO_4^{-1}	periodate	-1	BO_3^{-3}	borate	-3
IO_3^{-1}	iodate	-1	$\text{C}_6\text{H}_5\text{O}_7^{-3}$	citrate	-3
MnO_4^{-1}	permanganate	-1	PO_4^{-3}	phosphate	-3
NO_3^{-1}	nitrate	-1	PO_3^{-3}	phosphite	-3
NO_2^{-1}	nitrite	-1	$\text{Fe}(\text{CN})_6^{-3}$	hexacyanoferrate (III)	-3
$\text{H}_2\text{PO}_4^{-1}$	dihydrogen phosphate	-1	$\text{Fe}(\text{CN})_6^{-4}$	hexacyanoferrate (II)	-4
HSO_4^{-1}	bisulfate	-1	$\text{P}_2\text{O}_7^{-4}$	pyrophosphate	-4
HSO_3^{-1}	bisulfite	-1			